



Specific Rotation of Alpha and Beta -D-glucose

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"A freshly prepared solution of α -D-glucose shows a specific rotation of $+112^\circ$. Over time, the rotation of this solution decreases and reaches an equilibrium value of $+52.5^\circ$. In contrast, a freshly prepared solution of β -D-glucose has a specific rotation of $+19^\circ$. The rotation of this solution increases over time to the same equilibrium value of $+52.5^\circ$.

- Why do the specific rotations of the freshly prepared samples change over time, and why do they reach the same equilibrium value?
- Calculate the percentage of each anomer at equilibrium.

"

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